SYNTHESIS OF [³²P] LABELLED 1-O-ALKYL-2-DESOXY-2-AMINO-SN-GLYCERO-3-PHOSPHOCHOLINES

H. P. DEIGNER* and B. FYRNYS

Pharmazeutisch-Chemisches Institut, University of Heidelberg, Im Neuenheimer Feld 364, 69120 Heidelberg/Germany

SUMMARY

The syntheses of N-substituted 1-O-alkyl-2-desoxy-2-amino-sn-glycero-3- $[^{32}P]$ phosphocholines were performed in four steps starting from $[^{32}P]$ POCl₃ and the corresponding 1-O-alkyl-2-amino-propane-3-ols in 5-7 % total yield.

KEY WORDS

[³²P]etherphospholipids; 1-O-alkyl-2-desoxy-2-amino-sn-glycero-3-[³²P]phosphocholines

INTRODUCTION

Derivatives of 1-O-alkyl-2-desoxy-2-amino-sn-glycero-3-phosphocholine are potent inhibitors of phospholipase A_2 [1, 2] and potential ligands of the receptor of the platelet activating factor (PAF). Isotopically labelled, 2-amino-etherphospholipids therefore are of potential value in investigations involving interactions of phospholipids with cellular membranes as well as in metabolic studies.

In this report we describe the preparation of $[^{32}P]$ labelled 2-desoxy-2-amino-*sn*-glycero-etherphospholipids utilizing $[^{32}P]$ POCl₃ to introduce the radioactive label and 4-substituted 2-chloro-2-oxo-1,3,2-oxaza- $[^{32}P]$ phospholanes as versatile intermediates.

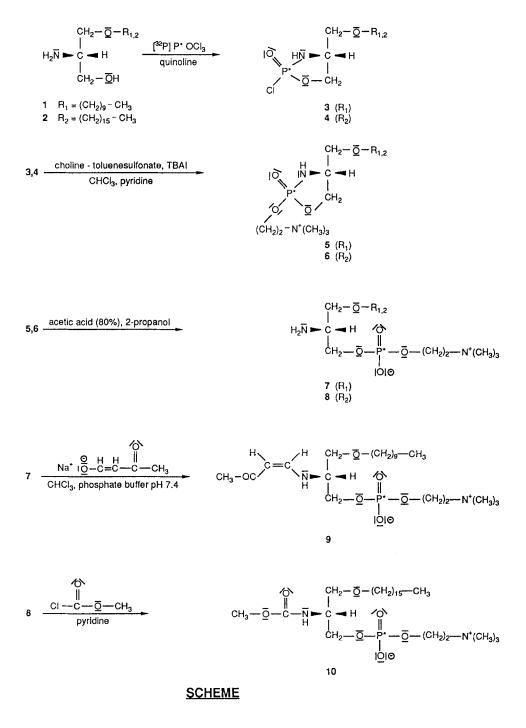
RESULTS AND DISCUSSION

The synthetic route to labelled 1-O-alkyl-2-desoxy-2-amino-sn-glycero-3-phosphocholines is outlined in the scheme. Preparation of $[^{32}P]$ labelled 2-desoxy-2-amino-lysophospholipids was carried out according to the synthetic sequence for unlabelled 2-amino-phospholipids published by us previously [3].

0362-4803/94/020185-05\$07.50 ©1994 by John Wiley & Sons, Ltd. Received 21 September, 1993 Revised 4 October, 1993

⁻⁻⁻⁻⁻⁻

^{*} to whom correspondence should be adressed



Equilibration of $[^{32}P]$ phosphoric acid with phosphorus oxychloride yielded isotopically labelled POCl₃; reaction with the corresponding 1-O-alkyl-2-amino-propane-3-ol (1, 2) gave the labelled oxazaphospholane intermediates (3, 4). The choline group was introduced by nucleophilic exchange of the chlorine, acid

hydrolysis of (5), (6) opened the ring to give the desired 1-O-alkyl-2-desoxy-2-amino-sn-glycero-[^{32}P]-phosphocholines (7) and (8). The total yield of the three consecutive steps was lower starting from the long-chain aminoalcohol (2) (12% vs. 32%) suggesting a decreased reactivity along with increasing carbon chain length. Conversion of the 2-amino-lysophospholipids (7) and (8) with the sodium salt of 1-hydroxy-but-1-en-3-on or with methylchloroformate and purification by chromatography on thin layer plates provided the labelled vinylogous amide (9) and the carbamic-acid-methylester (10) with a specific activity of approximately 720 MBq/mmol and a radiochemical purity of 95-96%. The specific radioactivity of the phospholipids was 0.78 of the theoretical value, a result which can be explained by incomplete exchange with [^{32}P]phosphoric acid.

MATERIALS

Tetrabutyl ammonium iodide was purchased from Fluka (Buchs, Switzerland), phosphorus oxychloride and p-toluenesulfonic acid from Aldrich (Steinheim, Germany). Chloroform and methanol were obtained from J. T. Baker (Deventer, Holland) and distilled from P_2O_5 and from Mg prior to use. Silica gel (grade 60, 70-230 mesh) for column chromatography and 2-propanol was from Machery-Nagel (Düren, Germany). TLC plates (0.5 mm, F 254) were from E. Merck (Darmstadt, Germany) and preeluted with methanol. [³²P]phosphoric acid (314-337 TBq/mmol) was from Du Pont de Nemours Deutschland GmbH (Bad Homburg).

METHODS

Thin-layer chromatography (TLC) and column chromatography were performed using a mixture of chloroform/methanol/water (65:45:8, v/v) as mobile phase. Phospholipids were detected with "Phospray" (Supelco, Bad Homburg, Germany). Mass spectra were obtained using a MAT 311 A mass spectrometer (Varian, Bremen, Germany) equipped with a FAB ion gun (Xe, 6 KV, ion current 1 mA) from Ion Tech (Teddington, U. K.) and glycerol as matrix. ¹H-NMR-spectra were recorded at 250 MHz (WM-250, Bruker Physik AG, Karlsruhe, Germany); tetramethyl silane was used as internal reference. Spectra were run in acetonitrile-D₃ or in CDCl₃/methanol-D₄, 2:1 (v/v). Multiplicities are reported as singlet (s), doublet (d), triplet (t) or multiplet (m).

[³²P] phosphorus oxychloride

Labelled phosphorus oxychloride was obtained applying the method of Keenan et al. [4].

 $[^{32}P]$ phosphoric acid (370 MBq, 314-337 TBq/mmol) was mixed with freshly distilled phosphorus oxychloride (30 µl, 0.32 mmol) to give a theoretical specific activity of 1.12 GBq/mmol (calculated for 100% conversion to POCl₃) and stirred for 24 h at 107°C in a screwed vial.

1-O-decyl-2-desoxy-2-amino-sn-glycero-[³²P]phosphocholine (7) and 1-O-hexadecyl-2-desoxy-2amino-sn-glycero-3-[³²P]phosphocholine (8)

To 12 μ l (129 μ mol) of [³²P] phosphorus oxychloride dissolved in chloroform, a solution of 30 mg (130 μ mol) of 1-O-decyl-2-amino-propane-3-ol (1) or 41 mg (130 μ mol) 1-O-hexadecyl-2-amino-propane-3-ol (2) and 35 μ l (290 μ mol) quinoline was added dropwise at 4°C. The mixture was allowed to warm up at room temperature, then stirred at 55°C for 16 h.

The resulting solution of 1-O-decyl-2,3-(2'chloro-2'oxo-1',3',2')-oxaza[³²P]phospholane (3) or 1-O-hexadecyl-2,3-(2'chloro-2'oxo-1',3',2')-oxaza[³²P]phospholane (4) was cooled to 12°C and 55mg (0.2 mmol) choline tosylate in 0.3 ml pyridine was added. After stirring for 24 h at 55°C, the solvents were removed *in vacuo*, and the residue containing the crude oxazaphospholanes (5) or (6) was redissolved in 1 ml of 2-propanol/acetic acid (80%) 3:2 (v/v). Hydrolysis was carried out by stirring at 50°C for 30 min and at room temperature for additional 2 h. The solvents were removed by distillation *in vacuo* at under 40°C and silica gel chromatography afforded 16.3 mg (41 μ mol, 31%) of (7) and 7.2 mg (15 μ mol, 12%) of (8); MS (FAB; glycerol; pos. mode): m/z =397 [M+H]⁺ (7) and m/z =481 [M+H]⁺ (8).

1-O-decyl-2-desoxy-2-(1'-amino-but-1'-en-3'-on)-sn-glycero-3-phosphocholine (9)

To 16.3 mg (41 μ mol) of (7), dissolved in 2 ml of a biphasic mixture (1:1, v/v) of chloroform/phosphate buffer (20 mM, pH 7.4) 200 mg (1.85 mmol) of the sodium salt of 1-hydroxy-but-1-en-3-on [5] were added and the solution stirred for 24 h at room temperature. The solvents were removed by distillation *in vacuo* and the residue purified by thin layer chromatography yielding 2.8 mg (6 μ mol, 15%) of (9) (cis configuration, specific activity 724 MBq/mmol)

MS (FAB; glycerol; pos. mode): $m/z = 465 [M+H]^+$

¹H-NMR (CDCl₃/D₄-methanol, 2:1) δ ppm:

6.8 (1H, d, $-C\underline{H}=CH-C=O$), 4.95 (1H, d, $-CH=C\underline{H}-C=O$), 4.2 (1H, m, $sn-2-C\underline{H}$), 3.95 (2H, m, $C\underline{H}_2-CH_2-N^+(CH_3)_3$), 3.9-3.8 (2H, t, $sn-3-C\underline{H}_2$), 3.6 (2H, m, $-C\underline{H}_2-N^+(CH_3)_3$), 3.45 (2H, m, $sn-1-C\underline{H}_2-$), 3.3 (3H, s, $O=C-C\underline{H}_3$), 3.2 (9H, s, $-N^+(C\underline{H}_3)_3$), 1.5 (2H, m, $-C\underline{H}_2-(CH_2)_8-CH_3$), 1.25 (16H, m, $-C\underline{H}_2-(C\underline{H}_2)_8-CH_3$), 0.85 (3H, t, $-(C\underline{H}_2)_9-C\underline{H}_3$).

1-O-hexadecyl-2-desoxy-2-amino-carbamic-acid-methylester-sn-glycero-3-phosphocholine (10)

To a solution of 7.2 mg (15 μ mol) (8) in chloroform/pyridine (3 ml, 5:1), 10 μ l (130 μ mol) of methylchloroformate were added dropwise at 0°C; the reaction mixture was allowed to warm up to room temperature and stirred for another 4 h. Evaporation *in vacuo* afforded crude (10) which was subjected to thin layer chromatography to give 4.8 mg (9 μ mol, 60%) of (10) (specific activity 715 MBq/mmol) MS (FAB; glycerol; pos. mode): m/z = 523 [M+H]⁺.

¹H-NMR (CD₃-CΞN) δ ppm:

4.2 (1H, m, sn-2-C<u>H</u>), 3.9 (2H, m, C<u>H</u>₂-CH₂-N⁺(CH₃)₃, 3.9-3.8 (2H, m, sn-3-C<u>H</u>₂), 3.6 (3H, s, -O-C<u>H</u>₃), 3.55 (2H, m, -C<u>H</u>₂-N⁺(CH₃)₃, 3.45 (2H, m, sn-1-C<u>H</u>₂-), 3.2 (9H, s, -N⁺C<u>H</u>₃)₃, 1.5 (2H, m, -C<u>H</u>₂-(CH₂)₁₄-CH₃), 1.25 (28H, m, -CH₂-(C<u>H</u>₂)₁₄-CH₃), 0.85 (3H, t, -(CH₂)₁₅-C<u>H</u>₃).

ACKNOWLEDGEMENT

The authors are indebted to Professor R. Neidlein, Pharmazeutisch-Chemisches Institut Heidelberg, for continous support. Support of this work by the Deutsche Forschungsgemeinschaft (DFG) is gratefully acknowledged.

REFERENCES

- 1. Davidson, F.F., Hajdu, J. and Dennis, E.A. Biochem. Biophys. Res. Comm. 137: 587 (1986).
- van den Berg, L., Franken, P.A., Verheij, H. M., Dijkman, R. and de Haas, G.H. Biochim. Biophys. Acta <u>1124</u>: 66 (1992).
- 3. Deigner, H.-P. and Fyrnys, B. Chem. and Physics of Lipids 61: 199 (1992).
- 4. Keenan, R.W, Martinez, R. A. and Williams, R. F. J. Biol. Chem. 257: 14817 (1982).
- 5. Johnson, Woroch, Mathews Am. Soc. 69: 570 (1947).